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THE APPLICATION OF THE NEW TYPE GRIGNARD REAGENT DERIVED FROM ETHYL  $\alpha$ -(tert-Butylsulfinyl)ACETATE TO THE SYNTHESIS OF  $\alpha$ , $\beta$ -UNSATURATED ESTERS.

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Recently, we have reported that a new type Grignard reagent derived from ethyl  $\alpha$ -phenylsulfinylacetate and alkylmagnesium halides has a good reactivity at the  $\alpha$ -carbon atom toward aldehydes and ketones to give the corresponding addition products.<sup>1)</sup> In this communication, we wish to report an application of the Grignard reagent derived from ethyl  $\alpha$ -(tert-butylsulfinyl)acetate(1) to the synthesis of  $\alpha,\beta$ -unsaturated esters.

When 1 was treated with 1 equiv of an alkylmagnesium halides in ethyl ether or benzene at 0°C under nitrogen, the Grignard reagent 2 was produced as a milky precipitate. The following experiment is typical of the procedure employed for the preparation of  $\alpha,\beta$ -unsaturated esters.<sup>2</sup>

To a solution of 2 derived from 583 mg(3 mmol) of 1 and ethylmagnesium bromide(3 mmol) in 3 ml of ether was added 1 equiv of n-butyraldehyde at 0°C. The stirring was continued for 12 hr at room temperature. The reaction mixture was neutralized with a dilute solution of tartaric acid and extracted with chloroform. The extract was washed, dried and evaporated to leave 674 mg of crude needles of 3a (84% yield), which was dissolved in 8 ml of carbon tetrachloride and treated with 413 mg(3 mmol) of sulfuryl chloride for 10 minutes at room temperature according to the method developed by Durst et al.<sup>3)</sup> The solvent was evaporated keeping the temperature below 40°C. The residual oil was passed through 15 g of alumina(200 mesh) with 30 ml of methylene dichloride to give 297 mg of (E)-ethyl 2-hexenoate(4a) (82% yield based on 3a).

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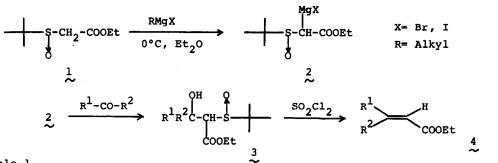
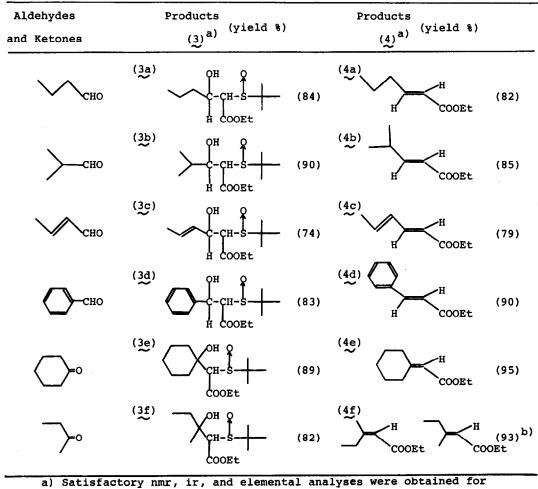


Table I.
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all new compounds described here. b) The E/Z ratio was ca. 1.

Similarly, isobutyraldehyde, crotonaldehyde, benzaldehyde, cyclohexanone and ethyl methyl ketone readily react with 2 to yield the corresponding  $\alpha,\beta$ -unsaturared esters (4b-f). The results are summarized in Table 1.

It is noteworthy that the  $\alpha,\beta$ -unsaturated esters 4a-d (except 4e,f) obtained from this reaction possess the E-configuration. However, when ethyl methyl ketone was subjected to the same reaction, the Z- $\alpha,\beta$ -unsaturated ester was produced together with the E-ester(E : Z = ca. 1 : 1).<sup>4</sup>

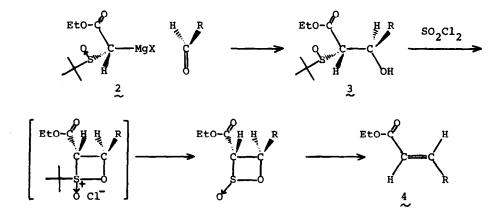
The assignment of E- and Z-configurations was based on nmr spectral data and gas chromatographical behavior. Further study on the application for organic syntheses of high stereospecificity in this reaction is in progress.

## Referrences and Notes

1) N. Kunieda, J. Nokami, and M. Kinoshita, Tetrahedron Lett., 3997(1974).

- The synthesis of α,β-unsaturated esters by the alkylative elimination of the anion of methyl α-phenylsulfinylacetate has been reported. [B. M. Trost, W. P. Conway, P. E. Strege, and T. J. Dietsche, J. Amer. Chem. Soc., <u>96</u>, 7165(1974); B. M. Trost and T. N. Salzmann, ibid., <u>95</u>, 6840(1973)].
- 3) F. Jung, N. K. Sharma, and T. Durst, J. Amer. Chem. Soc., 95, 3420(1973). They have reported that the E/Z ratio of the cis-elimination is generally not greater than 7 : 3.
- 4) Supposing the addition of the Grignard reagent (2) toward aldehydes obeys the model prediction proposed by Cram and Prelog<sup>5)</sup> in terms of the eclipsing effect of substituents at the reaction center, and further supposing the elimination of the adducts by treating with sulfuryl chloride  $(3 \rightarrow 4)$ proceeds through the stereospecific cis-elimination via a  $\beta$ -sultine proposed by Durst et al.,<sup>3)</sup> a plausible stereochemical course for the reaction to yield the E- $\alpha$ , $\beta$ -unsaturated esters preferentially is formulated as follows (Scheme 1), since the relative steric, bulk of the substituents on the  $\alpha$ -carbon atom of 2 is considered to be  $-H \langle -COOEt \langle -SO-Bu^{t}$ .

## Scheme 1



5) J. D. Morrison and H. S. Mosher, "Asymmetric Organic Reactions", Prentice-Hall, Inc., Englewood Cliffs, New Jersey (1971), and references therein.